

Practice Practice Practice...

Write the chemical formula for each of the following:

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|-----------------------|--------------------------|---------------------------------|------------------------|
| 1. Silver Chloride | 2. Copper(II) iodide | 3. Zinc sulphate | 4. ammonia |
| 5. Strontium Fluoride | 6. Potassium cyanide | 7. Aluminum nitrate | 8. Mercury(I) sulphide |
| 9. Manganese dioxide | 10. Potassium chromate | 11. Potassium nitrate | 12. silver hydroxide |
| 13. Lead(II) nitrate | 14. Iron (III) hydroxide | 15. Oxygen gas | 16. Sulphur trioxide |
| 17. Aluminum nitrite | 18. Sodium hydride | 19. Tin (II) hydrogen carbonate | 20. Iron(III) sulphate |

Write the correct name for each of the following:

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|--|-------------------------------------|---------------------------------------|--|
| 1. AgI | 2. CCl ₄ | 3. CuSO ₄ | 4. MgO |
| 5. NaCN | 6. PCl ₅ | 7. Fe(NO ₃) ₂ | 8. (NH ₄) ₂ SO ₄ |
| 9. SO ₂ | 10. Cu ₂ SO ₄ | 11. Fe(NO ₃) ₂ | 12. NaBr |
| 13. K ₂ C ₂ O ₄ | 14. RbSCN | 15. AlPO ₄ | 16. K ₂ CO ₃ |
| 17. Na ₃ PO ₄ | 18. SnSO ₄ | 19. Na ₂ CrO ₄ | 20. Pb(NO ₂) ₂ |

Write the chemical formula for each of the following:

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|--------------------------------|-------------------------|-------------------------|-------------------------------|
| 21. Sodium thiocyanate | 22. Cesium perbromate | 23. Zinc fluoride | 24. Barium hydrogen carbonate |
| 25. Tin (IV) sulphate | 26. Strontium bromate | 27. Hydrogen peroxide | 28. Nitrogen dioxide |
| 29. Aluminum hydrogen sulphite | 30. Potassium chlorite | 31. Zinc phosphate | 32. Lead (IV) iodite |
| 33. Ammonium perchlorate | 34. Chromium (VI) oxide | 35. Mercury (I) cyanide | 36. Calcium hypobromite |
| 37. Aluminum oxide | 38. Calcium sulphide | 39. Iron (III) iodine | 40. Magnesium hydroxide |

Write the correct name for each of the following:

- | | | | |
|--|--------------------------------------|---|---|
| 21. SnSO ₄ | 22. WF ₆ | 23. SO ₃ | 24. K ₂ Cr ₂ O ₇ |
| 25. Pb(C ₂ O ₄) | 26. Na ₂ HPO ₄ | 27. Al ₂ (CO ₃) ₃ | 28. Au(CN) ₃ |
| 29. KMnO ₄ | 30. SrH ₂ | 31. KHCO ₃ | 32. Al ₂ S ₃ |
| 33. Cu(C ₂ H ₃ O ₂) ₂ | 34. Ba(SCN) ₂ | 35. NaClO ₃ | 36. Fe(OH) ₂ |
| 37. NaBrO | 38. HNO ₂ | 39. V ₂ S ₅ | 40. Fe(ClO) ₂ |